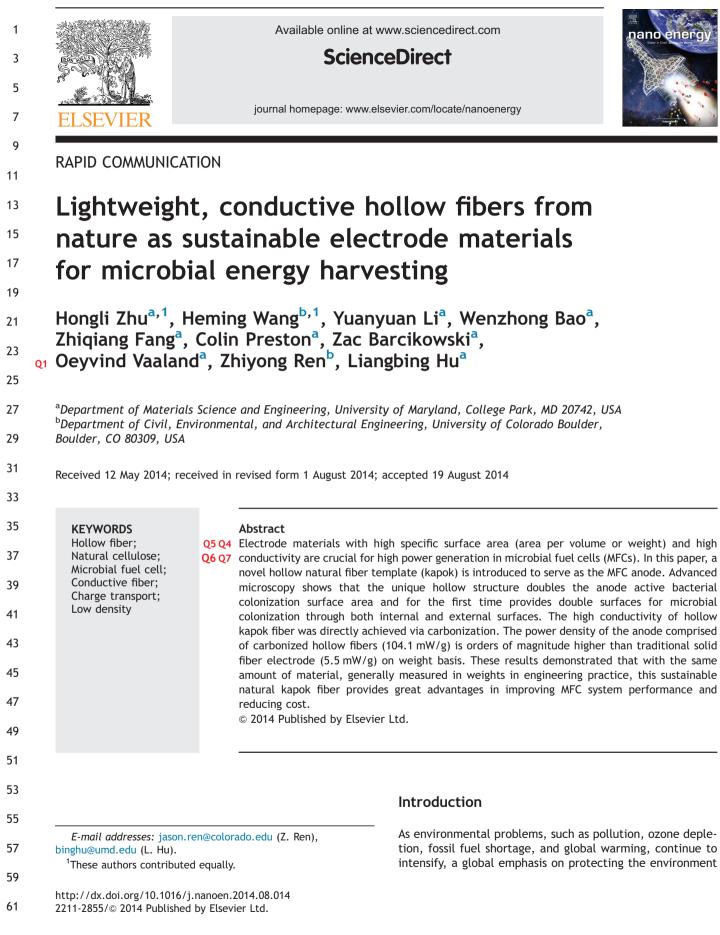
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becomes increasingly urgent. Wood is an abundant and carbon neutral renewable resource. Using renewable biomaterials is an efficient way to maintain a clean and sustainable world. The kapok [Ceibapentandra (L.) Gaertn] trees supply an abundant guantity of round micro-tubular fibers with a thin cell wall thickness of 0.8-1.0 μ m, a fiber diameter of 10-20 μ m, and a length of several centimeters [1]. These natural organic tubes are widely available, lowcost, environmentally friendly, light-weight, and sustainable. Various tubes with a high length to diameter ratio have been successfully fabricated with kapok fibers as the template [2,3].

13 A microbial fuel cell (MFC) converts organic matter to electricity using electrochemically active bacteria (EAB) as 15 the biocatalyst. A prospective application for MFCs is towards wastewater treatment, where the bacteria directly 17 breaks down organic matter in wastewater and simultaneously generates clean energy. The extracellular electron transfer between EAB and anode surface occurs through 19 direct contacts via pili and cytochromes or indirect electron 21 transfer via mediators [4-9]. An ideal anode electrode should have a high surface area, high conductivity, biocom-23 patibility, chemical stability, and three-dimensional (3D) macroporous structure to allow microbes to colonize or 25 access for electron transfer [10-17]. Various conductive porous and three-dimensional materials have been investi-27 gated as MFC anodes, including carbon cloth [18], conductive textiles [10], and carbon nanotube (CNT)/graphene 29 coated sponges [12,19], layered corrugated carbon [20], electrospun and solution blown carbon fibers [21], and 31 porous ceramic anode [22] however, all the reported fibers are made of materials from graphite mines and have a solid 33 structure with limited surface area that are difficult to be implemented in engineering systems due to the high weight, 35 high cost, and unsustainable nature. Only a few current studies have developed sustainable and low cost anodes 37 from natural materials for MFCs. Open structured macroporous bioanode was fabricated using a natural loofah 39 sponge as the precursor material. Carbonization and polyaniline modification of the loofah sponge greatly increased 41 its bacterial loading capacity and electron transfer, which favored much higher power production compared to traditional three-dimensional anodes [23]. Biochar manufactured 43 with one-step pyrolysis process of lodgepole pine sawdust 45 pellets or lodgepole pine woodchips was tested as a new anode material in two-chamber MFCs. The results show 47 satisfactory power outputs and it is 90% cheaper than granular activated carbon and graphite granules [24], but biochar anodes are still formed solid architecture which 49 limits their specific surface area. Another study prepared an 51 ordered 3D macroporous carbon anode for MFCs from the stem of kenaf (Hibiscus cannabinus), a natural plant; 53 however, many channels in the structure are blocked by valves with only the open channels available for biofilm 55 propagation [25].

The hollow fibers as electrode materials are distinctive from other porous and three-dimensional electrodes because all other electrodes are made of solid fibers that electrons can only travel on the outside surface, but the hollow structure fibers can provide both outside and inside surfaces for electron transfer. On the other hand, these hollow fibers are isolated from the natural plant Kapok which makes the

63 hollow fiber material biocompatible, sustainable and low cost. In this study, we carbonized a low-cost and naturally 65 forming hollow kapok fiber and used it as anode for the MFC. which provides a promising alternative to solid macroporous electrodes. The hollow structure of this conductive natural 67 fiber doubles the active bacterial loading area with no increase in volume and weight, which offers larger surface 69 areas for EAB to colonize and efficiently deliver electrons 71 through both external and internal walls. The power generation with conductive kapok fiber anode from artificial waste-73 water was compared with traditional carbon cloth anode in the single-chamber air-cathode MFCs. The introduction of 75 this high-performance anode from natural, low-cost, and sustainable electrodes for simultaneous waste management and renewable energy production provides a clean, cost-77 effective, and natural approach for sustainable community 79 development.

Experiment

Preparation of conductive kapok anode

Natural kapok fibers were purchased from Bamboo Fiber (Bamboo Fiber Corp., US). They were washed thoroughly with ethanol and distilled water, then a free standing and porous kapok paper was prepared by vacuum filtration. The kapok paper was carbonized in a mixed gas atmosphere with 95% Ar and 5% H_2 in a tube furnace (Applied Test Systems, Inc.) from room temperature to 400 °C with a ramp rate of 30 °C/h, and further increased the temperature to 1100 °C 93 with a ramp rate of 200 °C/h, then last held the samples at 1100 °C for 2 h. After carbonization, the fiber was dispersed in distilled water and re-filtered to form a circular film with 40 mm diameter. The control sample without carbonization was fabricated directly from natural kapok fiber via vacuum filtration. 2.0 mg CNT was pre-filtered as an electrode current collector. The thickness of the Kapok sample is \sim 100 μ m. 101

MFC construction and operation

Single chamber cubic-shaped MFCs were constructed as 105 previously described [26]. The empty volume of each MFC was 28 mL. Air cathodes (projected area of 4.5 cm²) made 107 by applying one carbon base layer, four PTFE diffusion layers and one Pt/C (0.5 mg cm^{-2}) catalyst layer on the 109 30% wet-proofed carbon cloth (Fuel Cell Earth LLC, MA, USA) were used in all the experiments [27]. The carbonized 111 kapok anodes (kapok_c) and non-carbonized kapok anodes (kapok_nc) were prepared as described above. Plain carbon 113 cloth (CC, Fuel Cell Earth LLC, MA, USA), commonly used as anode material, was used as a control sample for kapok 115 anodes

117 MFCs were inoculated using anaerobic sludge obtained from Littleton/Englewood Wastewater Treatment Plant (Englewood, CO). Medium solution was prepared containing 119 1.24 g L^{-1} CH₃COONa, 0.31 g L^{-1} NH₄Cl, 0.13 g L^{-1} KCl, 2.452 g L⁻¹ NaH₂PO₄ · H₂O, 4.576 g L⁻¹ Na₂HPO₄, 12.5 mL L⁻¹ 121 mineral solution, and 5 mLL^{-1} vitamin solution [28]. All MFCs were operated in fed-batch mode at room tem-123 perature. Fresh medium solution was refilled when the

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voltage dropped below 10 mV with an external resistor of 1000 $\Omega.$

5 Analyses

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The MFC voltages and electrode potentials were recorded over 7 10 min intervals using a data acquisition system (Keithley Instrument, OH, US). The anode potentials and cathode ٥ potentials were measured against a Ag/AgCl reference electrode (RE-5B, Bioanalysis) at a distance of 1 cm to the anode in 11 the reactor. Polarization curves were obtained by Linear Sweep Voltammetry (LSV) at a scan rate of 0.1 mV/s using a 13 potentiostat (PC4/300, Gamry Instruments, NJ). Electrochemical Impedance Spectroscopy (EIS) was conducted using the 15 same potentiostat to measure the internal resistance at the scan range from 10^5 Hz to 0.005 Hz with a small sinusoidal 17 perturbation of +10 mV. In both LSV and EIS tests, the anode was used as the working electrode and the cathode was used 19 as the counter electrode and reference electrode. Chemical oxygen demand (COD) was measured using a standard colori-21 metric method (Hach Company, CO).

COD removal efficiency is the ratio of total COD removed at the end of the batch to influent COD in percentage. The output power (*P*) of MFC was calculated by P = UI, where *U* is the voltage across the MFC anode and cathode and *I* is the MFC output current. Current density was normalized by the projected area of the anode (4.5 cm²). Power density was normalized by the projected area, reactor volume (28 mL) and the mass of the anode.

Non-carbonized and carbonized kapok and carbon cloth 31 samples were characterized with Hitachi SU-70 FESEM field effect scanning electron microscopy (SEM), performed using 33 a Jeol JXA 840A system (Jeol Instruments, Tokyo, Japan) running at 5-10 keV. Samples were fixed overnight by 35 Karnovsky's fixative (Electron Microscopy Sciences, CA, USA) at 4 °C, washed three times in a pH 7.2 phosphate 37 buffer, and then dehydrated stepwise in a series of ethanol/ water solutions with increasing ethanol concentration of 39 50%, 70%, 80%, 90%, and 100%. Samples were then kept in a desiccator prior to carbon coating and SEM observation. The 41 fiber surface area was measured with a Micromeritics TriStar II 3020 Porosimeter Test Station. The surface of Kapok fiber 43 after carbonization is $820 \text{ m}^2/\text{g}$.

Results and discussions

Kapok trees as shown in Figure 1(a) are widespread in 49 tropical areas. They are 60-70 m tall and their trunk expands to 3 m in diameter with palm like leaves. Adult 51 kapok trees produce several hundred seed pods with black seeds and fibers that are shades of yellow Figure 1(b). The 53 fluffy light fibers extracted from the seed pods of the kapok trees are used in pillows, life preservers, sleeping bags, and 55 insulation. These fibers were also recently investigated as an oil absorbent and sound absorber [29-32]. This study took 57 hollow kapok fibers as an anode in the MFC. High surface area is critical to increase the microbial loading and 59 improve the power density for large scale MFCs [33,34]. Due to the open 3D hollow structure, the microbes colonize 61 both the inner and outer surfaces of the fiber wall, as illustrated in Figure 1(c). The biofilm mass loading is doubled within the same electrode volume. The generated electrons transport along both the internal and external surfaces of the fiber wall, thus the double surfaces for microbial colonization dramatically increase the power density. Meanwhile, since the kapok fiber is produced in nature, the fiber has good biocompatibility, which benefits the contact between microbes and the anode to enhance the electron transfer efficiency. These natural, lightweight, and cost efficient hollow fibers are promising electrodes for MFCs.

Carbonization of biomass is a promising strategy for carbon materials from sustainable resources [35]. The kapok fiber is a natural biocomposite composed mainly of cellulose and lignin. Figure 2(a) and (b) shows the fiber morphologies before carbonization. Figure 2(a) illustrates the kapok fiber has a round shape and smooth surface with a diameter of 10-20 μ m. In Figure 2(b), the thin wall of kapok fibers is well defined. The fiber has a large empty lumen, which provides enough space to host the microbial growth and substrate reservation within the lumen. Since 77 v% of the fiber is hollow, the density of kapok fiber is as low as 0.384 g/cm^3 [36,37], which is much lighter than the solid fiber; for example, the density of carbon cloth is 1.75 g/cm^3 [38]. Weight is a huge concern when designing portable systems for truck delivery, because it can be very heavy when filled with water. The lightweight of conductive fibers in MFCs will reduce the cost in transportation and realize large-scale practical applications. The lightweight of kapok is therefore a great benefit for fuel cells of any size.

Kapok fibers function as a microbial growth scaffold and 93 more importantly facilitate the electron transfer in the MFC. The conductivity of kapok fiber is thus important, but the pristine fibers are not conductive. The threat of an energy 95 shortage continues to drive the pursuit of a facile, simple, 97 and environmentally friendly method to prepare advanced functional materials. There are a lot of studies on making 99 conductive carbonaceous materials from earth abundant biomass via carbonization [35,39]. We carbonized kapok fiber in a mixed gas, see Experiment section. The pyrolysis process 101 of cellulose mainly includes dehydration and depolymeriza-103 tion. The reason for using a slow ramp rate before 400 °C is to reduce the carbohydrate material burning loss thereby improving yield and properties of obtained carbon fiber 105 [39]. Figure 2(c and d) shows the fiber morphology after 107 carbonization. From the images we can see the hollow structure is well preserved after carbonization, and the fiber walls become thinner. More SEM images of kapok fiber after 109 carbonization are presented in Supplemental material Figure S1. In order to measure the fiber conductivity, we fabricated 111 two-probe devices with our hollow carbon fiber on glass 113 substrates. Uniform fibers were fixed and contacted by silver paste (Figure 2(e)) and the electrical characterization was carried out in a vacuum using a shielded probe station. We 115 observed a linear I-V curve with a source drain bias V_{ds} reaching $\sim 1 \text{ V}$ (Figure 2(f)). Such ohmic I_{ds} - V_{ds} behavior 117 indicates that the Schottky barriers at the contacts can be 119 excluded. The more fiber morphology and I-V curve are presented in Supplemental material (Figures S2 and S3). The volume resistivity of $\rho \sim 140 \pm 30 \,\Omega \,\mu m$ represents the 121 upper limit due to the contribution from contact resistance, from which we can see the kapok fiber has good conductivity 123 after carbonization.

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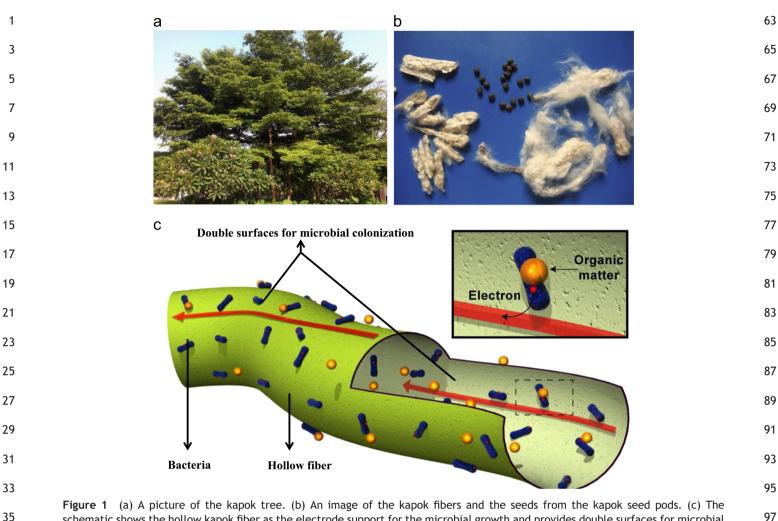


Figure 1 (a) A picture of the kapok tree. (b) An image of the kapok fibers and the seeds from the kapok seed pods. (c) The schematic shows the hollow kapok fiber as the electrode support for the microbial growth and provides double surfaces for microbial colonization through both internal and external surfaces. The kapok fiber applied in the MFC is post-carbonization with an excellent conductivity. Note that bacteria can grow both inside and outside the hollow fiber, maximizing the direct contact of bacteria with conductive fiber surface.

After acclimation, repeatable cycles of voltage genera-41 tion were obtained from all the reactors as shown in Figure 3(a). The voltage generation over a $1 k\Omega$ external 43 resistor in the kapok_c reactor is 514.9 ± 3.9 mV, which is 9.3% higher than that in the kapok_nc reactor 45 $(470.9\pm4.8 \text{ mV})$ and comparable to that in the CC reactor 47 $(520.0\pm2.6 \text{ mV})$. COD removal efficiencies were comparable in all the reactors, with $92.9 \pm 2.1\%$ for kapok_c, 92.5+1.0% for kapok nc, and 94.7+0.8% for CC. Under 49 stable performance, polarization curves were obtained by LSV and then power densities were calculated (Figure 3(b)). 51 The maximum power density generated in the kapok_c reactor and normalized by the projected anode surface 53 area is 1738.1 mW/m², which is 30.9% higher than that in the kapok nc reactor (1328.1 mW/m^2) and comparable to 55 that in the CC reactor (1689.8 mW/m^2). When normalized by reactor volume, the power densities of the kapok_c 57 reactor (27.9 W/m³) and the CC reactor (27.1 W/m³) are comparable, which are 27.2% higher than that of the 59 kapok_nc reactor (21.3 W/m³). Due to the hollow structure 61 of kapok fiber, the mass of the kapok anodes (kapok_c and kapok_nc) is only 7.5 mg, which is much lighter than the CC

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anode (139.1 mg). Normalizing the power densities by mass. 103 kapok_c and kapok_nc generated powers of 104.1 mW/g and 79.7 mW/g, respectively, but CC only produced 105 5.5 mW/g (Figure 3(c)). It should be noted that the power generation of kapok_nc MFC was due to the contribution of 107 CNTs on the electrode. CNTs are mixed with the kapok fiber in order to improve the mechanical strength of the kapok 109 film; otherwise it is hard to create a structure to serve as the electrode. Carbonization of non-conductive kapok fibers 111 indeed improves the performance of the MFC, with power densities increase by 30.9%, 31.0%, and 30.6%, respectively, 113 if normalized by electrode projected surface area, reactor volume and electrode mass. Figure 3(d) shows the anode 115 potentials and cathode potentials measured during LSV tests. All the cathode potentials are apparently similar at 117 each current density in the three reactors; the anode potentials are also similar in the kapok c reactor and the 119 CC reactor, but the anode potentials in the kapok_nc reactor are higher than that in the other two reactors. This 121 result confirms that the differences of power generation in the three reactors are due to different anode performances 123 but not cathode. Figure 3(e) shows the total resistance

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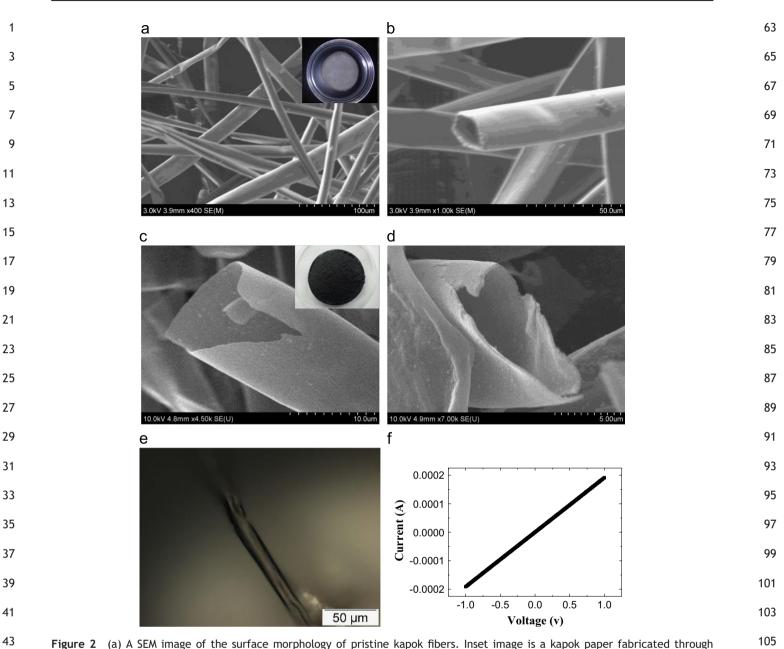


Figure 2 (a) A SEM image of the surface morphology of pristine kapok fibers. Inset image is a kapok paper fabricated through papermaking process. (b) A SEM image of the natural hollow tube. Both (a) and (b) are before carbonization. (c and d) SEM image of the kapok fiber morphology after carbonization. Inset picture in (c) is an image of kapok paper after carbonization. (e) A microscope image of a uniform single fiber fixed and contacted by silver paste for the conductivity test. (f) The *I-V* curve of a single kapok fiber in (e).

measured by an EIS test for MFCs with three different anode 51 materials. The intersection of the curve with the x-axis indicates the ohmic resistance, and the diameter of the 53 semicircle presents polarization resistance or charge transfer resistance [40]. The ohmic resistances of the reactors 55 with kapok c or kapok nc are about 72 Ω and 68 Ω , respectively, which are a little higher than that of CC (65 Ω). Because of the same reactor configuration and 57 electrolyte solutions, the ohmic resistances of all the 59 reactors are generally comparable. The ohmic resistances of the reactor with kapok electrode are a little higher than 61 the reactor with CC electrode which may be due to the relatively low conductivity of uncarbonized kapok materials

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and the loose connection between carbonized kapok electrode and the titanium wire which is used to link with the 113 outside circuit. The polarization resistances of kapok_c and CC are both very small; while the polarization resis-115 tance of kapok_nc was as high as 20Ω . Charge-transfer resistance is closely relevant to the properties of electrode 117 surface which determines the kinetics of electron transfer. Both CC and kapok c electrodes have small charge-transfer 119 resistances which can well assist electron transfer, however, the high charge-transfer resistance of kapok_nc is detri-121 mental to electron transfer due to the high resistance on the surface of kapok_nc electrode. The total resis-123 tance of the reactor includes both ohmic resistance and

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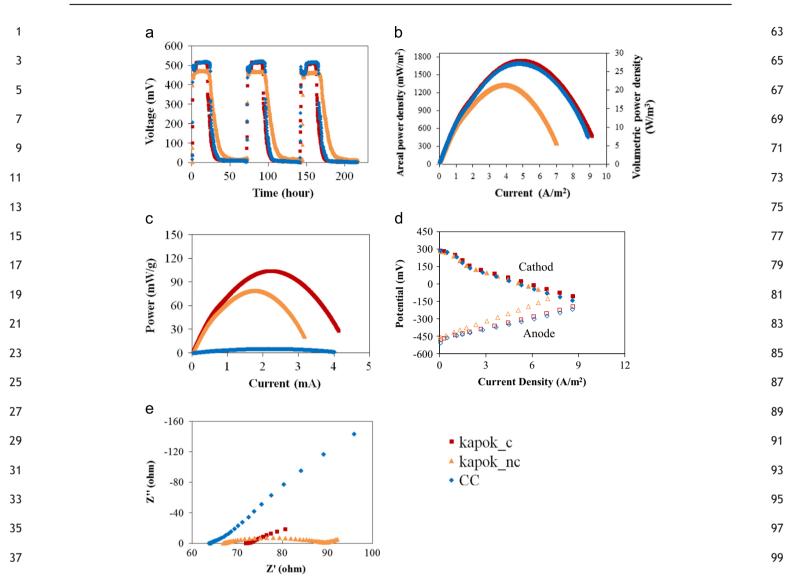


Figure 3 A demonstration of energy production in MFCs using carbonized kapok hollow fibers (kapok_c) as the anode with comparisons to uncarbonized kapok hollow fibers (kapok_nc) and traditional carbon cloth anodes (CC). (a) Repeatable voltage generations under a 1 kΩ external resistor. (b) Power production curves obtained from linear sweep voltammetry (LSV) test clearly
 show the maximum power density at the peak. Power densities are normalized to the projected anode surface area and reactor volume. (c) Power densities are normalized to anode mass. (d) Anode and cathode potentials measured during LSV test.
 (e) Electrochemical impedance spectroscopy (EIS) shows the total resistance of MFC with three different anode materials.

charge-transfer resistance, which is correlated to the performance of MFC reactors. Power productions of the 49 reactors with CC or kapok c are comparable since their resistances are also similar. The reactor with the kapok_nc 51 anode generated the lowest power because the high 53 resistance enhances the energy loss and decreases the power output. Since the surface conductivity is low because there is no carbonization treatment for kapok nc, which is 55 not favorable for electron transfer. Although the MFC with kapok c anode has a higher total resistance than the MFC 57 with CC anode, kapok_c can generate, at least, comparable 59 power density as CC, which indicates that the hollow structure can utilize both the inside surface and outside 61 surface to accelerate electron transfer, but the solid fiber anode, such as CC, can only transfer electrons through the

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outside surface. The kapok hollow fiber anodes in this study
are the first try, and their manufacturing procedures have
not been optimized yet. Therefore more efficient and
conductive kapok hollow fibers can be manufactured, and
the power production may be further enhanced following an
optimized manufacturing process.111

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The hollow conductive fiber provides a dual microbialanode interface. The microbial growth is evident on the internal and external wall of the fiber, which not only increases the surface area and biofilm loading, but more importantly provides double surfaces for microbial colonization on the anode. Solid surfaces of carbon cloth fibers are incapable of exhibiting this behavior (Figure 4(e) and (f)). The thickness of the biofilm depends on the oxygen and substrate concentration [41]. The anode chamber of the Q8

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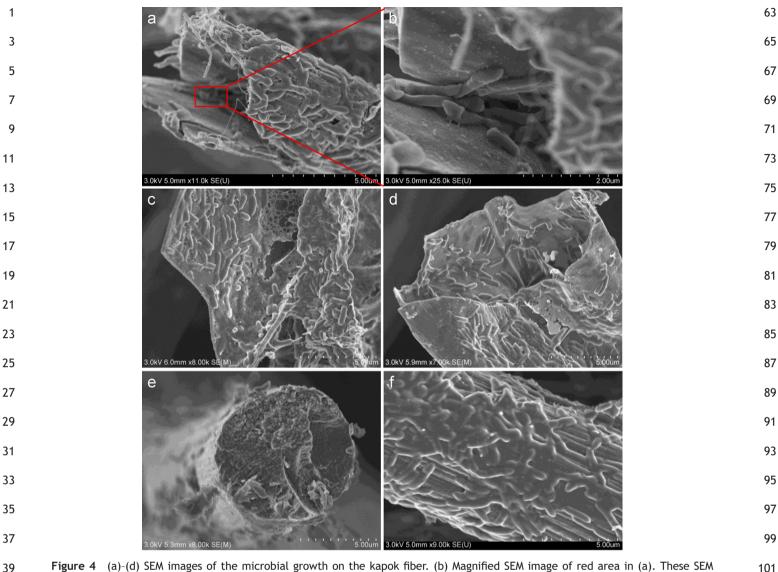


Figure 4 (a)-(d) SEM images of the microbial growth on the kapok fiber. (b) Magnified SEM image of red area in (a). These SEM 39 images clearly confirm that bacteria grow on both sides of the conductive hollow fiber wall. (e) and (f) SEM images of the microbial growth on the carbon cloth fiber. More SEM images of bacterial colonization on the kapok fiber and carbon cloth are presented in 41 103 Supplemental material Figure S4. 105

MFC reactors is in an anaerobic condition, which means the 45 biofim on the anode grows slowly in the limited oxygen environment. The COD of the substrate used in this study is 47 about 1000 mg/L, which is double that of typical domestic wastewater with a COD of 500 mg/L [42]. Biofilm cannot 49 grow as fast as that in the high concentrated substrate. SEM 51 images also clearly show that the diameter of the hollow fiber (μ m) is much bigger than bacteria sizes (0.5-1 μ m). In the three-month long lab testing of the hollow fiber anodes 53 in MFCs, no clogging was found and the performances of the reactors were stable. More studies are still needed to 55 evaluate long-term performances of the hollow kapok fiber. 09 For the solid carbon cloth fibers, the bacteria can only 57 colonize on the outer surface of solid carbon cloth fibers (Figure 4(e) and (f)), which limits the bacteria loading and 59 has a much larger density than the hollow fiber. Note that the mass density of carbonized kapok fibers in this study is 61 59.7 mg/cm³.

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Conclusion

This study introduces an open-structured, hollow 3D macro-109 porous anode formed from natural kapok fibers. The hollow structure doubles the microbe-anode interface while main-111 taining the same volume of solid fibers thus significantly increased the area for microbial acclimation and electron 113 transfer. The lightweight of such natural fiber provides a great advantage in large-scale engineering applications and 115 modular system development, as surface/weight ratio is a major challenge for scale-up now. The direct carbonization 117 of sustainable cellulosic fibers offers a facile and simple way to prepare conductive carbon fibers [43]. Normalizing with Q2 119 respect to the projected anode surface area and reactor volume, the power density production from the MFC using 121 this carbonized low cost hollow-fiber structure anode $(1738.1 \text{ mW/m}^2 \text{ or } 27.9 \text{ W/m}^3)$ is comparable to the expen-123 sive solid-fiber structured carbon cloth (1689.8 mW/m^2 or

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27.1 W/m³). Normalized power density by mass, the power density production using the lightweight carbonized kapok anode is 104.1 mW/g, which is about 20 times that produced by carbon cloth anodes (5.5 mW/g). The hollow natural fibers provide a new concept for designing high-power, lightweight, and sustainable MFC anodes with earth abundant and low cost materials that are greatly amenable to engineering applications.

Acknowledgment

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Appendix A. Supporting information

Supplementary data associated with this article can be found in the online version at http://dx.doi.org/10.1016/j.nanoen.2014.08.014.

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